SEASONAL VARIATION AND INTER-CORRELATION OF SO4²⁻, NO3⁻ NH4⁺ IN PM10 AEROSOLS AND TRACE GASES SO2 & NH3 AT HYDERABAD

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Simultaneous measurements of PM_{10} aerosols and trace gases were carried out at the urban site Hyderabad. These samples were analysed for $SO_4^{2^-}$, NO_3^- and NH_4^+ in water soluble extract. Simultaneously, trace gases SO_2 and NH_3 were also collected to understand their seasonal variation along with PM_{10} fractions of $SO_4^{2^-}$, NO_3^- and NH_4^+ . A brief description on methodology and results is given below-

Methodology

Samples of PM_{10} aerosols were collected at Hyderabad which lies in south-central India (17.5° N, 78.5° E) about 536m above sea level. The samples were collected at the terrace of the main building of our institute at a height of around 11m using Respirable Dust sampler (Envirotech APM 460 NL) on Whatman 41 filters on 24 hr basis with an average flow rate of 1.1 m³/min. These samples were stored in polythene bags and refrigerated. Trace gases (SO₂ and NH₃) were collected using gaseous attachment of Respirable Dust sampler (Envirotech APM 411 TE) using wet chemical method. The absorbing reagent used for SO₂ was a mixture of Na₂CO₃ and NaHCO₃ solution while dilute H₂SO₄ was used for NH₃. These gases were also collected on 24 hr basis with an approximate flow of 1 LPM.

To extract water soluble fraction of SO_4^{2-} , NO_3^- and NH_4^+ , a 2 x 2 cm² size portion was cut from the filter and transferred in to 7 ml of distilled deionized water. It was extracted using ultrasonic bath (Enertech Electronic Pvt.Ltd.) for 75 min. This extraction showed an efficiency of 99% for these components.

The analysis of $SO_4^{2^-}$, NO_3^- and NH_4^+ was carried out using an Ion Chromatography (Metrohm Basic 792). The samples of SO_2 were determined by Ion Chromatography (as $SO_4^{2^-}$) using anion column while the samples of NH_3 were determined (as NH_4^+) using cation column.

Results and Discussions:

Table 1 gives average concentration and standard deviation of $SO_4^{2^-}$, NO_3^- and NH_4^+ during winter and summer seasons of 2007. Samples collected during January and February months are considered as winter samples (n = 17) while during March-May are considered as summer season samples (n = 40).

It is to be noted that concentrations of SO_4^{2-} , NO_3^{-} and NH_4^{+} are higher during winter season as compared to summer season. This may be due to favourable meteorological conditions. During winters, the boundary layers comes down enhancing the density of pollutants in ambient air. Also, the increased humidity in winter season

catalyzes the oxidation of SO₂ and NO₂ which form SO₄²⁻, NO₃⁻ aerosols (Kulshrestha et al., 2005). It is observed that in PM₁₀, SO₄²⁻, NO₃⁻ aerosols are correlated with NH₄⁺ aerosols. As shown in Fig. 1 & 2, during winter, NH₄⁺ showed better correlation with SO₄²⁻ (r^2 = 0.67) and NO₃⁻ (0.63) as compared to summer season (Fig. 3 & 4) when r^2 values were 0.19 and 0.19 respectively. These observations indicated that presence of (NH₄)₂SO₄ and (NH₄)NO₃ aerosols is more prevalent during winter than in summer.

Table 1: Concentrations of SO_4^{2-} , NO_3^{-} and $NH_4^+(\mu g/m^3)$ in PM_{10} aerosols
during winter and summer seasons.

	WINTER		SUMMER	
	(N=17)		(N=40)	
	Average	SD	Average	SD
SO_4^{2-}	24.1	8.7	11.8	8.1
NO ₃ ⁻	13.0	3.6	11.9	4.7
$\mathrm{NH_4}^+$	15.1	5.7	7.0	3.0

Table 2 gives the concentrations of SO₂ and NH₃ during winter and summer seasons. It is seen from the table that there is no much seasonal difference for SO₂ but NH₃ showed slightly higher value during summer season. With subject to correlation analysis, data showed no significant correlation between SO₄²⁻ & SO₂ and NH₄⁺ & NH₃. Further details on these results will be discussed during the conference.

Table 2: Concentrations of SO_2 and NH_3 ($\mu g/m^3$) in gaseous samples during winter and summer seasons.

	WINTER (N=38)		SUMMER (N=68)	
	Average	SD	Average	SD
SO ₂	12.9	5.7	12.3	7.8
NH ₃	13.7	6.2	15.2	8.8



Fig. 1 $PM_{10} SO_4^{2-} vs NH_4^+$ during winter season Fig. 2 $PM_{10} NO_3^- vs NH_4^+$ during winter season





Fig. 3 $PM_{10} SO_4^{2-}$ vs NH_4^+ during summer season summer season

Fig. 4 PM₁₀ NO₃⁻ vs NH₄⁺ during

Acknowledgement:

We sincerely thank Dr J S Yadav, Director IICT, Hyderabd for his constant encouragements and ISRO-GBP for providing financial support.

References:

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